

# INTRODUCTION

- Water exceeding 100 mg/l as  $CaCO_3$  is unpleasant to drink.
- Scaling in industrial and domestic water pipes.
- Novel Electrochemical process to remove (Ca<sup>+2</sup>) and (Mg<sup>+2</sup>) from extremely hard water like (LRGV) tap water.

# WATER HARDNESS PROBLEM in LRGV



**US population** dependent on **Ground water** 

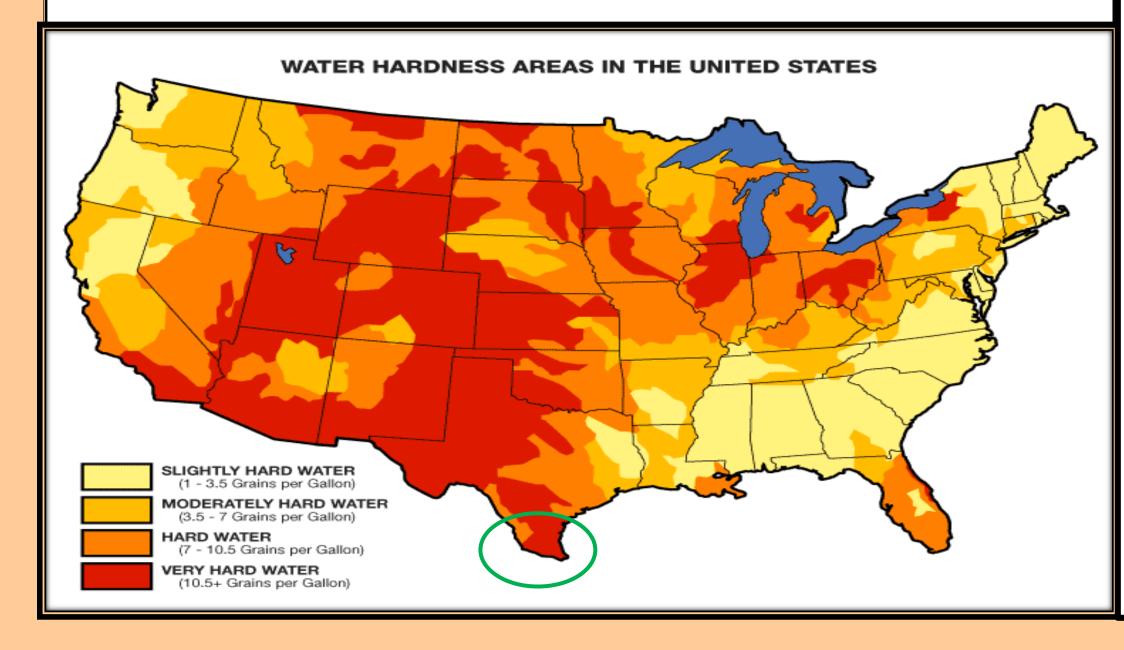


20% of total withdrawal nationwide, is



85% boilers have hardwater supply

<b>Existing Methods</b>	Cons
<ul> <li>Chemical</li> <li>Ion Exchange</li> <li>Reverse Osmosis</li> <li>Electrodialysis</li> </ul>	Costly Costly Heavy Chemical Usage
<ul> <li>Nano filtration</li> <li>Crystallization</li> <li>Distillation</li> <li>Evaporation</li> </ul>	Post Processing Sludge



# Hardness removal from LRGV tap water using Graphite-**Concrete Electrodes**

Mirza Addaito Billah(20544291), K I M Iqbal, Jongmin Kim, Phillip Park Department of Civil Engineering, UTRGV

# **ELECTROCHEMICAL PRECIPITATION**

• Cathodic Reactions:  $O_2 + 2H_2O + 4e^- \rightarrow 4OH$  $2H_2O + 2e^- \rightarrow H_2\uparrow + 2OH^-$ 

 Scaling reactions  $Ca^{2+} + HCO_3^- + 2OH^- \rightarrow CaCO^ _{3}\downarrow + H_{2}O$  $Mg^{2+}+2OH \rightarrow Mg(OH)_2$ 



## **METHODOLOGY**

### **Casting Specimen**

5,7.5, 10% graphite content by volume



# EXPERIMENTAL SETUP

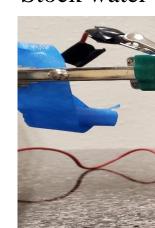
### **Power Supply:**

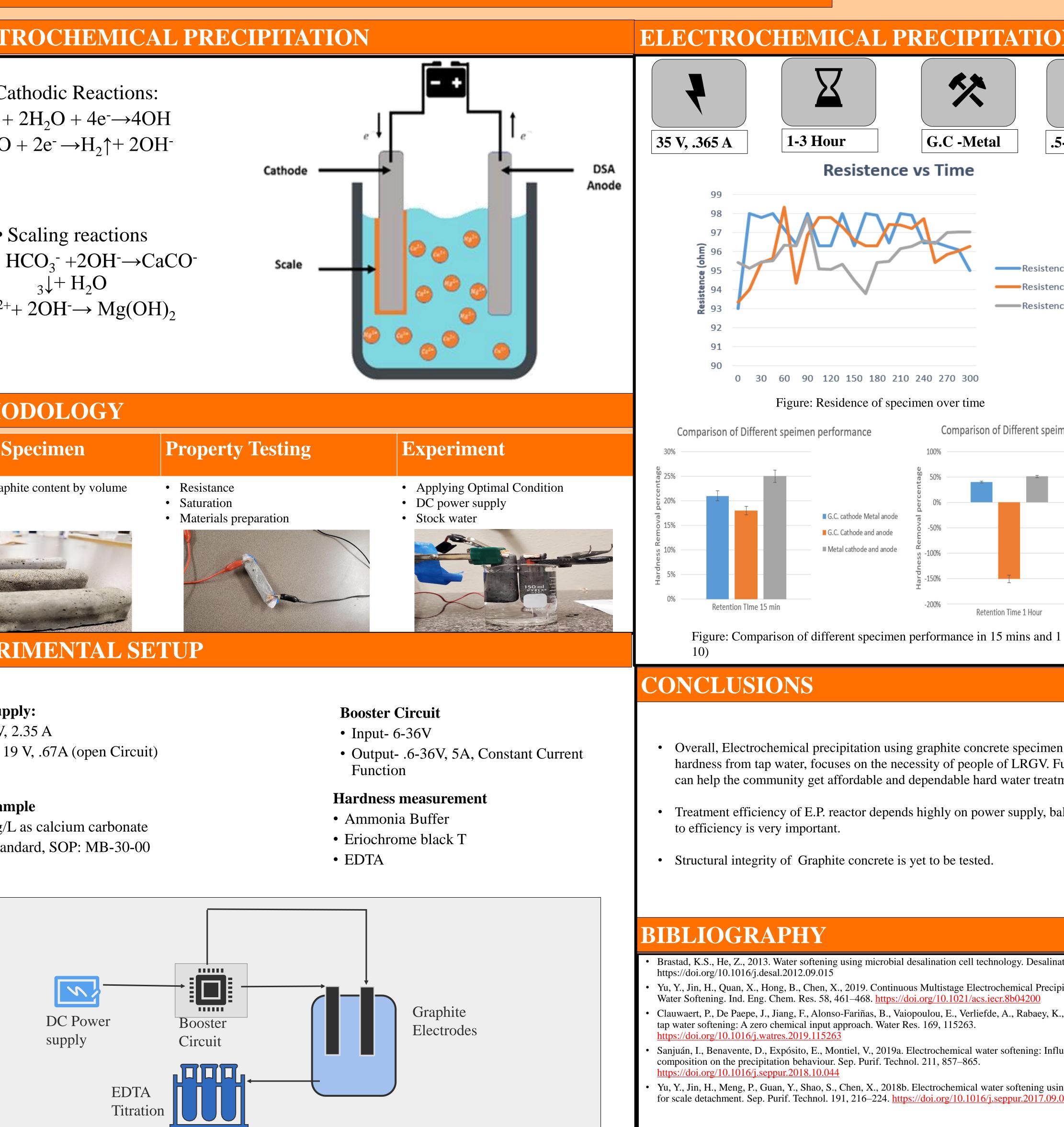
- 5 W,19V, 2.35 A
- Output- 19 V, .67A (open Circuit)

### Water sample

- 330 mg/L as calcium carbonate
- EPA Standard, SOP: MB-30-00









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<ul> <li>G.C. Cathode and anode</li> <li>Metal cathode and anode</li> </ul>
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