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Introduction

Energy production and storage have been on the forefront of research for the past decade. As fossil fuels are diminished prices are increasing making fossil fuels less and less attractive as an energy source. DSSC solar cells have shown much promise showing increased efficiency, increased output and lifetime. Porphyrins are an attractive sensitizer due to its strong absorption bands in the visible region, thermal stability and good absorption in the semiconductor surface.

Objectives:

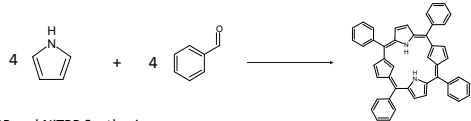
- To synthesize tetraphenylporphyrin (H₂TTP), Copper-tetraphenylporphyrin (CuTPP) and Nickel-tetraphenylporphyrin (NiTPP).
- Fabricate a dye-sensitized solar cell based on H₂TTP, CuTPP and NiTPP.
- Determine the open circuit voltage.

Methods

Preparation of dyes

H₂TTP Synthesis:

H₂TTP was synthesized by a modified version of the Adler method. Freshly distilled pyrrole was mixed in a 1:1 ratio with benzaldehyde in propionic acid as solvent and catalyst. The reaction mixture was refluxed at 145 for 30 minutes. The reaction mixture was then cooled to room temperature. After cooling the reaction was further cooled in an ice bath. The resulting deep-purple crystals were collected by vacuum filtration and washed with cold methanol and air dried.



CuTPP and NiTPP Synthesis:

H₂TTP was mixed with a metal chloride in a 1:1 ratio in DMF and microwave assisted synthesis. The reaction was performed using reflux at 153°C for 5 minutes. The samples were cooled naturally to room temperature, filtered and dried.

Preparation of TiO₂ photoanodes

10 g of TiO₂ nanopowder was dispersed into 88 g of ethanolic solution (98% ethanol/water) by using ultrasonication for 30 min. 1.8 g of TIPP was added dropwise. Then the mixture was stirred for 5 hours. TiO₂ were dip-coated on cleaned FTO glass four times and dried in air for 30 minutes, and then heat treated at 120 °C for 1 hour.

The counter electrode was prepared through the deposition of soot from a burning candle.

Preparation of the electrolyte

1.08 g of KI and 0.2 g of I₂ were dissolved in 20 mL of acetonitrile and 5 mL of ethylene glycol and stirred for 30 minutes.

XRD Characterization

- A Bruker D2 Phaser, equipped with a Cobalt source, with a step of 0.05°, and a 5 s counting time. From 20 to 60 in 2 theta.

FTIR Characterization

- Perkin Elmer Frontier MIR / UTAR (Universal Total Attenuated Reflection)
- Collection range: 600 cm⁻¹ to 4000 cm⁻¹
- Resolution of 4.0 cm⁻¹

UV-VIS Characterization

- Perkin Elmer Lambda 950 UV-VIS/NIR Spectrometer
- Collection range: 800 cm⁻¹ to 250 cm⁻¹
- Resolution of 1 nm

Open circuit Voltage

- 10031S Southwire Multimeter.

Results

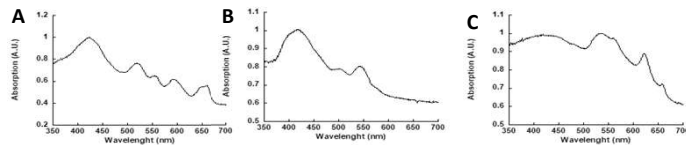


Figure 1: UV-Vis spectrum for the A.) H₂TTP, B.) CuTPP C.) NiTPP,

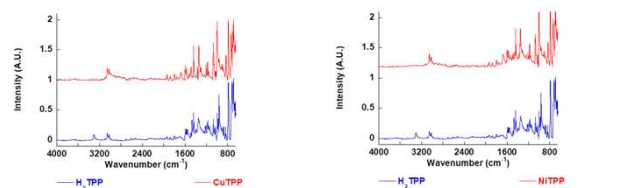


Figure 2: FTIR of the H₂TTP and CuTPP compounds synthesized under microwave assisted conditions.

Figure 3: FTIR of the H₂TTP and NiTPP compounds synthesized under microwave assisted conditions.

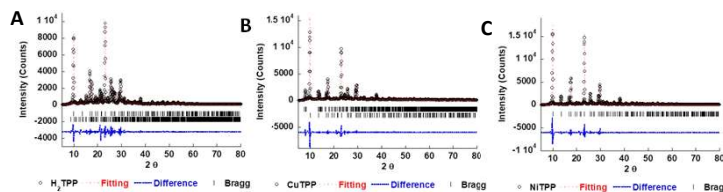


Figure 4: Powder X-ray diffraction of the A.) H₂TTP, B.) CuTPP and C.) NiTPP.

Sample	Space Group	a(Å)	b(Å)	c(Å)	α(°)	β(°)	γ(°)	GOF
H ₂ TTP	I-42/d	15.20	15.20	13.95	90.00	90.00	90.00	4.1
	P212121	12.02	19.30	14.69	90.00	90.00	90.00	
CuTPP	P21/M	15.09	8.64	12.00	90.00	116.02	90.00	3.5
	I-42/d	15.10	15.10	13.99	90.00	90.00	90.00	
NiTPP	I-42/d	15.09	15.09	13.89	90.00	90.00	90.00	4.3
TiO ₂	I41/amd	3.78	3.78	9.51	90.00	90.00	90.00	3.47

Table 1: Fullprof fitting results for the H₂TTP, CuTPP, NiTPP and PdTPP microwave assisted synthesized compounds

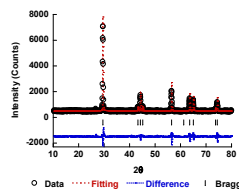


Figure 5: Diffraction patterns of TiO₂.

DSSC Semiconductor-Sensitizer material	Voltage(V)
TiO ₂	0.088
TiO ₂ -H ₂ TTP	0.175
TiO ₂ -CuTPP	0.258
TiO ₂ -NiTPP	0.052

Table 2: DSSC open circuit voltage with H₂TTP, CuTPP and NiTPP sensitizer.

Discussion

Synthesis

- Anatase was successfully synthesized sodium sulfate and reflux method under oxygenation at 100°C. The XRD patterns were consistent with the literature.
- Anatase Rutile mixture 80:20% was successfully synthesized sodium sulfate and reflux method under oxygenation at 80°.

Characterization:

- Diffraction patterns were consistent with the literature
- Copper TPP and H₂TTP show two crystal phases present
- Nickel TPP Diffraction pattern showed only one phase
- TiO₂ was determined to be in the anatase phase

Binding Of Porphyrin to TiO₂

- Solubility of TPP compounds was low in selected solvent and impregnation of the solid TiO₂ with H₂TTP, CuTPP, and NiTPP were low (only faint color was observed in the Fixed TiO₂)

Cell Functionality

- The cells functioned well under ambient inhouse light
- Porphyrin modified cells with exception of NiTPP showed enhanced voltage over TiO₂ alone

Future work

TiO₂ in mixture of Anatase/Rutile
Increase H₂TTP, CuTPP, NiTPP in TiO₂
Purify the TPP before metalation

- single phase TPP
- Determination of the following parameter of the DSSC:
 - Fill Factor
 - Current Density
 - Efficiency

References

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Acknowledgements

